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### (54) Title: CYTOKINE PRODUCTION INHIBITORS

#### (57) Abstract

A 5,6-dihydro-α-pyrone of formula (I) wherein R is CO2H or CH3 and each of R1 and R2 is H; or R is CO2H, one of R1 and R2 is H and the other is OH; or when R is CO<sub>2</sub>H, a pharmaceutically veterinarily acceptable Processes for salt thereof. compounds producing formula (I) and their use as cytokine production inhibitors are also described.

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## CYTOKINE PRODUCTION INHIBITORS

The present invention relates to 5,6-dihydro-α-pyrones useful as cytokine production inhibitors, to the preparation of these compounds and to pharmaceutical and veterinary compositions containing them.

We have now discovered that fermentation of a strain of the fungus *Phomopsis* sp. in a nutrient medium produces two 5,6-dihydro-α-pyrones esterified in the 5-position with an unsaturated C<sub>14</sub> fatty acid and also the free C<sub>14</sub> fatty acid. We have also discovered that fermentation of a strain of the fungus *Paecilomyces* sp. in a nutrient medium produces a useful phomalactone.

The present invention therefore provides a 5,6-dihydro- $\alpha$ -15 pyrone of formula (I):

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wherein R is  $CO_2H$  or  $CH_3$  and each of  $R^1$  and  $R^2$  is H; or R is  $CO_2H$ , one of  $R^1$  and  $R^2$  is H and the other is OH; or, when R is  $CO_2H$ , a pharmaceutically or veterinarily acceptable salt thereof.

Preferred compounds of the invention are:

3-((5S,6S)-5,6-dihydro-5-((6S)-4,6-dimethyldodeca-2E,4E-dienoyl)-2H-pyran-2-on-6-yl)-prop-2E-enoic acid; and

3-((5S,6S)-5,6-dihydro-5-((6S)-4,6-dimethyldodeca-2E,4E-dienoyl)-2H-pyran-2-on-6-yl)-prop-2E-ene.

The present invention provides a process for the preparation of a 5,6-dihydro-α-pyrone of formula (I) or a pharmaceutically or veterinarily acceptable salt thereof, which process comprises:

- (i) fermenting, in a source of carbon, nitrogen and inorganic salts, fungal strain *Phomopsis* sp. 22502 (CBS 313.96) or a mutant thereof which produces a said 5.6-dihydro- $\alpha$ -pyrone;
- (ii) isolating a said 5,6-dihydro- $\alpha$ -pyrone from the fermentation broth; and

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(iii) if desired, when the isolated 5,6-dihydro- $\alpha$ -pyrone is the compound of formula (I) wherein R is  $CO_2H$ , converting the said 5,6-dihydro- $\alpha$ -pyrone into a pharmaceutically or veterinarily acceptable salt thereof.

The compounds of formula (I) have been isolated from a microorganism which we have designated X22502 and which has been identified as a strain of the genus *Phomopsis* (Saccardo)

20 Bubák on the basis of the following morphological data with reference to the description given by SUTTON, B.C., 1980 (The Coelomycetes. Farnham Royal: Commonwealth Agricultural Bureaux):

The fungal strain Phomopsis sp. (X22502) (CBS 313.96) is

25 a coelomycete isolated from tropical freshwater foam which
produces a dense, dark grey-olivaceous (Flora of British Fungi
Colour Identification Chart, 1969, Edinburgh: HMSO) mycelium
with a white lobate margin at 24°C on 2% malt extract agar with
glucose and peptone (MEA: composition per litre of distilled

water: Difco malt extract, 20g; Bacto-peptone, 1g; agar, 20g).

After 7 days the mycelium attains a diameter of 2.5 - 3.5cm.

Conidiomatal development is stimulated by exposure to near-UV light. Conidiomata are solitary, carbonaceous, 5 unilocular, ostiolate and measure 1.5 - 2.0mm wide and 1.25 -2.0mm high. Conidiogenous cells are borne on branched conidiophores which line the conidiogenous cavity. These cells are hyaline, obclavate to cylindrical, integrated, phialidic and measure 16 - 20  $\mu m \times 1.5$  - 2.0 $\mu m$ . Conidia are hyaline, aseptate and generally of three types: A-conidia (5.5. .... -  $7.5\mu m \times 1.5$  -  $3.0\mu m$ ) are ellipsoid to fusiform, usually with acute apices and a guttule at each end; B-conidia (20 -  $32\mu m \times 10^{-3}$ <1 $\mu$ m) are hamate and filiform; C-conidia (9.5 - 11.5 $\mu$ m x 1.5 - $3.0\mu\mathrm{m})$  are obclavate with acute apices and usually at least 15 three guttules. All three conidial types can be found within a single conidioma. The fluvial origin and observed microscopical characters did not allow further classification to species.

The 5,6-dihydro-α-pyrones of formula (I) are associated primarily with the mycelium on termination of the fermentation. They may be recovered and purified from the medium. The separation and purification of the compounds from the fermentation broth and their recovery can be achieved using solvent extraction followed by chromatographic fractionation. The 5,6-dihydro-α-pyrone of formula (I) in which R is CO<sub>2</sub>H may be converted into pharmaceutically or veterinarily acceptable salts by conventional methods. Suitable salts include salts with alkali metals such as sodium or potassium and ammonium salts.

The 5,6-dihydro- $\alpha$ -pyrone of formula (I) wherein R is  $CH_3$  can, alternatively, be produced by the esterification of the phomalactone of formula (II):

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$$\begin{array}{c} H \\ HO \\ \hline \\ O \\ O \end{array} \qquad \text{(II)}$$

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with (6S)-4,6-dimethyldodeca-2E,4E-dienoic acid which is a fatty acid of formula (IIIa):

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Preferably the reaction is carried out in the presence of a dehydrating agent such as DCC (dicyclohexylcarbodiimide) or EDC (1-(3-dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride) and dimethylaminopyridine. The reaction is typically carried out in an inert solvent such as dichloromethane or tetrahydrofuran.

The reagents are generally mixed with stirring for example at a low temperature such as -78°C. The reaction is then allowed to warm to room temperature (20-25°C) and stirred until complete. The reaction may be monitored

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chromatographically by thin layer chromatography or reversed phase high performance liquid chromatography and is typically complete within sixteen hours. Other dehydrating agents such as an alkylchloroformate and triethylamine,

phenyldichlorophosphate, 2-chloro-1,3,5-trinitrobenzene and pyridine, and chlorosulphonyl isocyanate can also be used under similar conditions.

Alternatively an excess of the phomalactone of formula (II) can be reacted with the acid of formula (IIIa). The

water formed can be removed by azeotropic distillation.

Suitable solvents include toluene and 1, 4-dioxane. The reaction is typically catalysed by acids such as sulphuric acid and p-toluenesulphonic acid.

It is advantageous to produce the 5,6-dihydro- $\alpha$ -pyrone of formula (I) wherein R is  $CH_3$  by this route as both the phomalactone and the fatty acid can be produced by fermentation in larger quantities than the 5,6-dihydro- $\alpha$ -pyrone of formula (I) wherein R is  $CH_3$ .

The fatty acid of formula (IIIa) is one of a group of 20 fatty acids of the following formula (III):

wherein one of  $R^1$  and  $R^2$  is H and the other is H or OH. These fatty acids can be obtained by fermentation of the fungal strain *Phomopsis* sp 22502 (CBS 313.96) or a mutant thereof.

- 5 In accordance with the present invention, therefore, the fatty acid of formula (III) can be produced by a process which comprises:
  - (i) fermenting, in a source of carbon, nitrogen and inorganic salts, strain *Phomopsis* sp. 22502 (CBS 313.96) or a

mutant thereof which produces the said fatty acid; and

(ii) isolating the said fatty acid of formula (III) from the fermentation broth.

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The fatty acid of formula (III) may, of course, be isolated from the same fermentation broth as the 5,6-dihydro-15  $\alpha$ -pyrones of formula (I). The fatty acid, like the 5,6-dihydro- $\alpha$ -pyrones, is primarily associated with the mycelium on termination of fermentation.

Some of the fatty acids of formula (III) are novel. The present invention therefore further provides a fatty acid of 20 formula (IIIb):

wherein one of  $R^1$  and  $R^2$  is H and the other is OH

The phomalactone of formula (II) can be synthesised by methods known in the prior art, for example those disclosed in Krivobok, S. et al, Pharmazie, (1994): 49, H8, 605-607;

5 Guirand, P. et al, Pharmazie, (1994): 49, H8, 279-281;
Krasnoff, S.B. et al, J. Chem. Ecol., (1994): 20, 293-302 and
Murayama, T. et al, Agric. Biol. Chem., (1987): 51, 2055-2060.

The present invention does however provide a new process for the preparation of the phomalactone of formula (II), which 10 process comprises:

- (i) fermenting, in a source of carbon, nitrogen and inorganic salts, fungal strain Paecilomyces sp. 3527 (CBS 314.96) or a mutant thereof which produces the said phomalactone; and
- 15 (ii) isolating the said phomalactone from the fermentation broth.

The phomalactone is found primarily in the culture liquor on termination of the fermentation and may be recovered and purified. The separation and purification of the compound from the fermentation broth and its recovery can be achieved using solvent extraction followed by application of conventional chromatographic fractionations with various chromatographic techniques and solvent systems.

The phomalactone of formula (II) has been isolated from a microorganism which we have designated X3527 and which has been identified as a strain of the genus *Paecilomyces* Bainier on the basis of the following morphological data with reference to the descriptions given by SAMSON, R.A., 1974

(Paecilomyces and some allied Hyphomycetes. Studies in Mycology No. 6. Baarn: CBS). The fungal strain Paecilomyes sp. (X3527) (CBS 314.96) is an entomogenous hyphomycete isolated from a tropical Lepidoptera pupa which produces white mycelium attaining 4.5 - 5.0cm diameter within 14 days at 25°C on 2% MEA. The aerial mycelium becomes powdery as conidiogenesis occurs and may develop denser concentric zones. A pale-buff-yellow pigmentation frequently develops in aerial and/or submerged mycelium.

Conidiophores are hyaline and smooth-walled with stipe dimensions of 100 - 400μm x 1.5 - 2.5μm and bear single or sparsely clustered phialides. Phialides are produced with the characteristic morphology of Paecilomyces Bainier measuring 7 - 20μm long with a lower section inflated to 2 - 2.5μm wide.

The phialide neck is often considerably attenuated (<1μm wide) and bent. Phialides of very variable morphology are also produced, e.g. lacking an inflated basal region and/or with an attenuated neck measuring about 20μm long. Conidia are ellipsoid to cylindrical (3.5 - 7μm x 2 - 3μm), smooth-walled, hyaline and borne in conspicuous imbricate dry chains.

Further classification as a Paecilomyces anamorph of Cordyceps (Fries) Link may be justified on the grounds of the lepidopterous origin of the isolate coupled with the results of numerous studies made by H.C. Evans and R.A. Samson (unpublished data) of entomopathogenic Paecilomyces anamorphs of Cordyceps.

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The strains X22502 and X3527 were deposited by Xenova Group plc of 240 Bath Road, Slough, Berkshire, SL1 4EF, United

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Kingdom under the Budapest Treaty at the Centraalbureau voor Schimmelcultures, Baarn, the Netherlands, on 19th March 1996 under references X07/64/502 and X08/64/527 respectively.

Strain X22502 was assigned the reference number CBS 313.96.

5 Strain X3527 was assigned the reference number CBS 314.96.

The present invention also embraces mutants of the above microorganisms. For example, those which are obtained by natural selection or those produced by mutating agents including ionising radiation such as ultraviolet irradiation, or chemical mutagens such as nitrosoguanidine or the like treatments, are also included within the ambit of this invention.

The invention further provides a biologically pure culture of fungal strain X22502 or X3527 or of a mutant

15 thereof which produces the compounds of the invention. Such cultures are substantially free from other microorganisms.

The invention also provides a process for fermenting the fungal strain X22502, X3527 or a said mutant, which process comprises fermenting strain X22502 or X3527 or a said mutant

20 thereof in a source of carbon, nitrogen and inorganic salts.

Assimilable sources of carbon, nitrogen and minerals may be provided by either simple or complex nutrients. Sources of carbon will generally include glucose, maltose, starch, glycerol, molasses, dextrin, lactose, sucrose, fructose, carboxylic acids, amino acids, glycerides, alcohols, alkanes and vegetable oils. Sources of carbon will generally comprise from 0.5 to 10% by weight of the fermentation medium.

Sources of nitrogen will generally include soya bean meal, corn steep liquors, distillers' solubles, yeast

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extracts, cottonseed meal, peptones, ground nut meal, malt extract, molasses, casein, amino acid mixtures, ammonia (gas or solution), ammonium salts or nitrates. Urea and other amides may also be used. Sources of nitrogen will generally comprise from 0.1 to 10% by weight of the fermentation medium.

Nutrient mineral salts which may be incorporated into the culture medium include the generally used salts capable of yielding sodium, potassium, ammonium, iron, magnesium, zinc, nickel, cobalt, manganese, vanadium, chromium, calcium, copper, molybdenum, boron, phosphate, sulphate, chloride and carbonate ions.

An antifoam may be present to control excessive foaming and added at intervals as required.

Fermentation can be conducted at temperatures ranging

from 20°C to 40°C, preferably 24-30°C. For optimal results, it is most convenient to conduct these fermentations at a temperature in the range 24-26°C. The starting pH of the nutrient medium suitable for producing the compounds can vary from 5.0 to 8.5 with a preferred range of from 5.0 to 7.5.

Small scale fermentations are conveniently carried out by placing suitable quantities of nutrient medium in a flask by known sterile techniques, inoculating the flask with either spores or vegetative cellular growth of the fungal strain, loosely stoppering the flask with cotton wool, and permitting the fermentation to proceed in a constant room temperature of about 25°C on a rotary shaker at from 95 to 300 rpm for 2 to 10 days. The fermentation may also be conducted in static culture on liquid or semi-solid medium.

For larger scale work, it is preferable to conduct the

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fermentation in suitable tanks provided with an agitator and a means of aerating the fermentation medium. The nutrient medium is made up in the tank after sterilization and is inoculated with a source of vegetative cellular growth of the fungal strain. The fermentation is allowed to continue for from 1 to 8 days while agitating and/or aerating the nutrient medium at a temperature in the range 20°C to 37°C. The degree of aeration is dependent upon several factors such as the size of the fermenter and agitation speed. Generally the larger scale fermentations are agitated at about 95 to 750 rpm and aerations of about 0.5 to 1.5 VVM (volumes of air per volume of medium per minute).

The separation of the present compounds from the whole fermentation broth and their recovery is carried out by

15 solvent extraction followed by application of chromatographic fractionations with various chromatographic techniques and solvent systems. The present compounds in pure form have thus been isolated in this way.

The 5,6-dihydro- $\alpha$ -pyrones of formula (I) and pharmaceutically and veterinarily acceptable salts of the compound of formula (I) wherein R is  $CO_2H$  are inhibitors of the production of cytokines, specifically IL-1 $\beta$ .

These compounds can therefore be used in the treatment of disorders requiring immunosuppression, for example

25 immunoinflammatory conditions and CNS disorders. A human or animal, e.g. a mammal, can therefore be treated by a method comprising administration of a therapeutically effective amount of a compound of formula (I), or a pharmaceutically or veterinarily acceptable salt of the compound of formula (I)

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wherein R is CO<sub>2</sub>H.

These compounds can be used in the treatment of an immunoinflammatory condition such as rheumatoid arthritis, osteoarthritis, septic shock, psoriasis, atherosclerosis, inflammatory bowel disease, Crohn's disease and asthma. The compounds of the present invention also exhibit pharmacological properties associated with the treatment of other disorders requiring immunosuppression, for example central nervous system (CNS) disorders such as encephalomyelitis and Alzheimer's disease.

The compounds of the present invention can be administered in a variety of dosage forms, for example orally such as in the form of tablets, capsules, sugar- or film-coated tablets, liquid solutions or suspensions or parenterally, for example intramuscularly, intravenously or subcutaneously. The present compounds may therefore be given by injection or infusion.

The dosage depends on a variety of factors including the age, weight and condition of the patient and the route of administration. Typically, however, the dosage adopted for each route of administration for adult humans is 0.001 to 10 mg/kg, most commonly in the range of 0.01 to 5 mg/kg, body weight. Such a dosage may be given, for example, from 1 to 5 times daily orally or by bolus infusion, infusion over several hours and/or repeated administration.

The toxicity of the compounds of the invention is negligible, they can therefore safely be used in therapy.

The compounds of the present invention are formulated for use as a pharmaceutical or veterinary composition also

comprising a pharmaceutically or veterinarily acceptable carrier or diluent. The compositions are typically prepared following conventional methods and are administered in a pharmaceutically or veterinarily suitable form.

For example, the solid oral forms may contain, together with the active compound, diluents, such as lactose, dextrose, saccharose, cellulose, corn starch or potato starch; lubricants such as silica, talc, stearic acid, magnesium or calcium stearate and/or polyethylene glycols; binding agents such as starches, arabic gums, gelatin, methylcellulose, carboxymethylcellulose, or polyvinyl pyrrolidone; disintegrating agents such as starch, alginic acid, alginates or sodium starch glycolate; effervescing mixtures; dye-stuffs; sweeteners; wetting agents such as lecithin, polysorbates, laurylsulphates. Such preparations may be manufactured in known manner, for example by means of mixing, granulating, tabletting, sugar coating, or film coating processes.

Liquid dispersions for oral administration may be syrups, emulsions and suspensions. The syrups may contain as carrier,

20 for example, saccharose or saccharose with glycerol and/or mannitol and/or sorbitol. In particular a syrup for diabetic patients can contain as carriers only products, for example sorbitol, which do not metabolise to glucose or which only metabolise a very small amount to glucose. The suspensions

25 and the emulsion may contain as carrier, for example a natural gum, agar, sodium alginate, pectin, methylcellulose, carboxymethylcellulose or polyvinyl alcohol.

Suspensions or solutions for intramuscular injections may contain, together with the active compound, a pharmaceutically

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acceptable carrier such as sterile water, olive oil, ethyl oleate, glycols such as propylene glycol, and if desired, a suitable amount of lidocaine hydrochloride. Solutions for intravenous injection or infusion may contain a carrier, for example, sterile water which is generally Water for Injection. Preferably, however, they may take the form of a sterile, aqueous, isotonic saline solution. Alternatively, the compounds of the present invention may be encapsulated within liposomes.

The following examples illustrate the invention Example 1

### Batch Fermentation of Phomopsis sp. 22502

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A 1.5ml cryovial containing 1 ml of macerated vegetative mycelium suspended in a 10% glycerol solution was retrieved

15 from storage at -135°C. A preculture was produced by aseptically placing 1ml of starting material in a 250ml baffled Erlenmeyer flask containing 40ml of nutrient solution

S1 and shaken at 240 rpm for 3 days at 25°C.

An intermediate culture was generated by aseptically
transferring the preculture to 2L of nutrient solution S1 in a
3L fermenter. The fermenter was agitated at 500rpm, aerated
at 0.5vvm, and the temperature controlled at 25°C for 3 days.

A production culture was generated by aseptically transferring an intermediate culture to a 75L fermenter containing 50L of nutrient solution P1. The production fermenter was stirred at 350rpm, aerated at 0.5vvm, and temperature controlled at 25°C. After 5% days incubation the fermentation was stopped and the culture was harvested.

The nutrient solutions used were as follows, percentages

being by weight:

Nutrient Solution S1: 1.5% glycerol, 1.5% soya bean peptone,
1% D-glucose, 0.5% malt extract, 0.3% NaCl, 0.1% CaCO<sub>3</sub>, 0.1%:
Tween 80, 0.1% Junlon PW110[suppliers: Honeywell and Stein,

Sutton, Surrey, U.K.] adjusted to pH 6

Nutrient Solution P1: 3.6% molasses, 0.2% casein hydrolysate,

0.004% phytic acid, 0.09% calcium chloride, 0.1%: Tween 80,
adjusted to pH 5

### 10 Example 2

Extraction and purification of the 5,6-dihydro- $\alpha$ -pyrones of formula (I) wherein  $R^1$  and  $R^2$  are both H, and the fatty acid of formula (IIIa) from *Phomopsis* sp. 22502

The broth in Example 1 was harvested by filtration using

15 a Schenk Niro 430 filter press, the clarified filtrate was

discarded and the retained biomass was extracted with 25L of

recirculating methanol for 24 hours. The methanolic extract

was harvested via filtration through the filter press and

evaporated to an aqueous concentrate using a thin film

20 evaporator.

The aqueous concentrate (10L) was then back extracted with 2 x 7L of an ethyl acetate:hexane (1:1) mix. The solvent extracts were pooled and evaporated to a gum under reduced pressure and redissolved in 50ml of ethyl acetate:hexane

5 (1:1). Purification was achieved by normal phase chromatography using a Biotage Flash 75 chromatography system and a Flash 75 KP-Sil silica (32-62 μm 60 Å) column (internal diameter (ID) 7.5 x 30cm length) and an isocratic mobile phase

(ethyl acetate:hexane 1:1 mix, 200ml/min flow rate). 1L fractions were collected and analysed by thin layer chromatography using the same mobile phase as the developing solvent.

The 5,6-dihydro-α-pyrone of formula (I) wherein R is CO<sub>2</sub>H 5 (Rf 0.53), the 5,6-dihydro- $\alpha$ -pyrone of formula (I) wherein R is CH<sub>3</sub> (Rf 0.89) and the fatty acid of formula (IIIa) (Rf 0.75) rich fractions were pooled, evaporated to dryness under reduced pressure and subjected to further purification by 10 preparative reversed phase HPLC using a Waters NovaPak C18 (100 Å  $5\mu M$ ) column (ID 2.5 x 20 cm length) and an isocratic mobile phase (80% acetonitrile: 20% water plus 0.1% v/v glacial acetic acid, flow rate 50 ml/min). Wavelength monitoring was at 278nm. The peaks collected at 11-14 15 minutes, 19-21 minutes and 30-32 minutes were evaporated to dryness to yield the 5,6-dihydro- $\alpha$ -pyrone of formula (I) wherein R is CO<sub>2</sub>H (7.5g), the fatty acid of formula (IIIa) (0.35q) and the 5,6-dihydro- $\alpha$ -pyrone of formula (I) wherein R is CH<sub>3</sub> (0.9g), respectively.

20 Physicochemical data for the three compounds are set out in Tables 1 to 3 below. Tables 2 and 3 show <sup>1</sup>H and <sup>13</sup>C NMR assignments respectively.

Table 14

	5,6-dihydro-α-pyro	one of formula (I)	fatty Acid of
			formula
		<u> </u>	(IIIa)
	R is CO₂H	R is CH <sub>3</sub>	
DCI-MS (m/z)	391 (MH) ·	361 (MH) '	225 (MH)*
5 Molecular	C <sub>22</sub> H <sub>30</sub> O <sub>6</sub>	$C_{22}H_{32}O_4$	C <sub>14</sub> H <sub>24</sub> O <sub>2</sub>
formula			
UVλ <sub>max</sub> nm	206, 276	204,274	265
IR (KBr)ucm-1	3391, 2960, 2928,	3390, 2960, 2928,	2900, 2685,
	2855, 1717, 1620,	1720, 1615, 1560,	2589
	1570, 1396, 1285,	1250, 1180, 1010,	1687, 1618,
	1250, 1161, 1024,	980.	1459
	980.		1417, 1285,
			1207
			1028, 984,
			940,
	· · · · · · · · · · · · · · · · · · ·		852, 700

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Table 2

	†δH/ppm in MeOH-d4				
Position	5,6-dihydro-α-pyron				
	R is CO <sub>2</sub> H R is	СН3	Fatty Acid of formula (IIIa)		
2					
3	6.35 (1H, d, 9.7)	6.29 (1H, d, 9.8)			
4	7.21 (1H, dd, 9.7, 5.6)	7.18 (1H, dd, 9.8, 5.6)	È		
5	5.64 (1H, dd, 5.6, 3.0)	5.48 (1H, dd, 5.6, 3.1)			
6	5.49 (1H, m)	5.19 (1H, ddq, 7.2, 3.0, 0.9)			
. 7	6.91 (1H, dd, 15.7, 5.0)	5.72 (1H, ddq, 15.4, 7.2, 1.7)			
8	6.32 (1H, dd, 15.7, 1.8)	6.05 (1H, dqd, 15.4, 6.6, 1.1)			
9	. 15.7, 1.0)	1.84 (3H, ddd, 6.8, 1.7, 0.8)			
1'		1.7, 0.3,			
1'-OH			11.70 (1H, br s		
2'	5.88 (1H, d, 15.6)	5.95 (1H, dd, 15.7,	5.77 (1H, d,		
3 '	7.41 (1H, dd, 15.6, 0.6)	7.41 (1H, dd, 15.6, 0.8)	7.39 (1H, d,		
4'	15.0, 0.0,	0.0,	13.07		
5'	5.83 (1H, br d, 9.8)	5.83 (1H, br d, 9.2)	5.72 (1H, d, 9.8)		
6'	2.66 (1H, m)	2.65 (1H, m)	2.53 (1H, m)		
7'-11'	1.3-1.5 (10H, m)	1.4 (10H, m)	1.2-1.4 (10H, m		
12'	0.97 (3H, t, 6.9)	0.95 (3H, t, 6.9)	0.87 (3H, t,		
13'	1.87 (3H, d, 1.1)	1.81 (3H, d, 1.2)	6.7)		
14'	1.08 (3H, d, 6.6)	1.09 (3H, d, 6.6)	1.78 (3H, d, 0.8)		
			0.98 (3H, d, 6.6)		

<sup>45 \*</sup> values obtained in CDCl<sub>3</sub>

 $<sup>\</sup>dagger$  The J values are in parenthesis (Hz)

Table 3

Position	1	δC/ppm in MeOH-d4				
	5,6-dihydro-α-py	5,6-dihydro-α-pyrone of formula (I)				
	R is CO₂H	R is CH,	Fatty Acid of formula (IIIa)			
2	164.6	165.5				
3	125.6	126.1				
4 5	143.4	143.4				
5	64.9	65.6				
6	79.6	81.4				
7	139.6	125.5				
8	128.2	138.8				
9	170.7	18.2				
1'	168.0	168.0	173.1			
2 '	115.3	115.5	114.7			
3 '	153.5	153.2	152.1			
4 '	133.3	133.1	131.2			
5 '	151.7	151.5	149.8			
6'	35.0	34.8	33.2			
7'	38.7	38.6	37.1			
8 '	33.4	33.2	31.8			
9'	30.9	30.7	29.3			
10'	29.0	28.9	27.3			
11'	24.1	23.9	22.5			
12'	14.8	14.6	13.9			
13'	12.9	12.7	12.1			
14'	21.1	20.9	20.2			

<sup>\*</sup> values obtained in CDCl3

30 Extraction and purification of the 5,6-dihydro-α-pyrones of formula (I) wherein one of R<sup>1</sup> and R<sup>2</sup> is H and the other is OH, and the fatty acids of formula (IIIb) from Phomopsis sp 22502.

The title compounds all possess a hydroxy substituent (as R<sup>1</sup> or R<sup>2</sup>) and are therefore more polar analogues of the

35 compounds of formulae (I) and (IIIa) produced as described in Example 2.

The title compounds were isolated from the broth of Example 1 as minor fermentation components, using purification

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methods similar to those described in Example 2.

Physicochemical data for the three compounds are set out in the following Tables 4 to 6. Tables 5 and 6 show  $^1\mathrm{H}$  and  $^{13}\mathrm{C}$  NMR assignments, respectively.

Table 4

5

	Compound of fo	ormula (I)	Fatty acid of	formula (IIIb)
	A	B	<u>C</u>	D
	(R <sup>1</sup> =H, R <sup>2</sup> =OH)	(R <sup>1</sup> =OH, R <sup>2</sup> =H)	(R <sup>1</sup> =H, R <sup>2</sup> =OH)	(R1=OH, R2=H)
DCI-MS $(m/Z)$	424 (MNH <sub>4</sub> +)	424 (MNH <sub>4</sub> +)	258 (MNH <sub>4</sub> +)	258 (MNH <sub>4</sub> +)
	407 (MH+)	407 (MH+)	240 (MH+)	240 (MH+)
Molecular				
formula	C <sub>22</sub> H <sub>30</sub> O <sub>7</sub>	C22H30O7	C <sub>14</sub> H <sub>24</sub> O <sub>3</sub>	C14H24O3
UV A max nm	203, 275	205, 274	267	266
IR(KBr)v cm <sup>-1</sup>	3444, 2956,	3400, 2900,	3255, 2900,	3363, 1692,
	1723, 1621,	1714, 1620,	1694, 1627,	1622, 1285,
	1289, 1249,	1286, 1247,	1382, 1285	1198, 1029,
	1159, 1107,	1156, 1106,	1193, 982	983
	981	980		

Table 5

	Position		δH/ppm in MeOH-d4					
		A	B	C	D			
	2							
5	3	6.35 (1H,d,9.8)	6.35(1H,d,9.8)					
	4	7.21 (1H,dd,9.7,	7.21(1H,dd,					
	_	5.7) 5.67 (1H,dd,5.7,	9.8, 5.8) 5.67(1H,dd,6.3					
	5	3.0)	,5.3)					
10	6	5.52 (1H,m)	5.52 (1H,m)					
	7	6.99(1H,dd,15.7,	6.98(1H,dd,					
		4.5)	15.7, 4.7)					
	8	6.30(1H,dd,15.7,	6.31 (1H,dd,					
		1.8)	15.7, 1.9)					
15	9							
	1'							
	2,	5.87 (1H,d,15.5)	5.87 (1H,d,	5.87(1H,d,	5.85(1H,d,			
	~	3.07 (2.07 0, 0.07)	15.6)	(15.6)	15.6)			
20	3′	7.40 (1H,d,15.3)	7.40(1H,d,	7.38(1H,d,	7.40(1H,d,			
			15.5	15.9)	15.7)			
	4'							
	5′	5.84(1H, brd,	5.83(1H, brd,	5.78(1H, brd,	5.75(1H,d,			
25	6'	9.0) 2.67 (1H,m)	9.9) 2.67 (1H,m)	9.8) 2.68(1H,m)	9.8) 2.65(1H,m)			
25	7'	1.3-1.5 (8H,m)	1.3-1.55(8H,m)	1.35-1.6	1.4-1.6(8H,m)			
	,	1.5 1.5 (0.1//	2.0 2.00 (0.1,,	(8H,m)				
	8'	и . «	u	и	4			
	9'	3.50 (1H,m)	ii	3.52(1H,m)	4			
30	10'	1.3-1.5 (8H,m)	44	1.35-1.6(8H,	ű			
		,,		m) "				
	11'	7 00 (377 # 7 4)	3.77(1H,m)		3.78(1H,m)			
	12'	1.00 (3H,t,7.4)	1.23(3H,d,6.1)	1.00(3H,t, 7.4)	1.20(3H,d, 6.2)			
35	13'	1.87 (3H,s)	1.88(3h,d,1.0)	1.90(3H,d,	1.90(3H,s)			
-				0.8)	_			
	14'	1.09 (3H,d,6.6)	1.08(3H,d,6.6)	1.10(3H,d,	1.10(3H,d,			
				6.6)	6.6)			
					0			

Table 6

Position	δC/ppm in MeOH-d4			
	Ā	B	Ç	D
2	164.0	164.0		
3	125.2	125.2		l
4	142.8	142.8		
5 ·	64.2	64.2		
6	78.9	78.9	}	
7	141.2	141.2		
8	125.4	125.5		
9	168.7	168.8		
1'	167.5	167.6	171.0	171.0
2'	114.8	114.8	116.8	116.8
37	153.2	153.1	151.6	151.6
4'	132.9	132.9	132.9	132.8
5′	151.2	151.2	149.6	149.7
6′	34.6	34.6	34.5	34.4
7'	31.1	38.3	31.1	38.4
8'	38.4	28.9	38.5	28.7
9'	73.9	27.0	73.9	27.0
10'	38.0	40.2	38.0	40.2
11'	24.8	68.6	24.9	68.6
12'	12.5	23.5	12.6	23.6
13'	10.3	12.5	10.3	12.6
14'	20.7	20.7	20.8	20.9

30

Example 4
Batch Fermentation of Paecilomyces sp. 3527

Starting material of the strain *Paecilomyces* sp. 3527 was generated by suspending a mature slant culture, grown on MEA

35 (2% malt extract, 1.5% agar), in 5ml 10% aqueous glycerol.

1ml of this suspension, in a 1.5ml cryovial, comprises the starting material which was retrieved from storage at -135°C.

A preculture was produced by aseptically placing 1ml of

starting material in a 250ml baffled Erlenmeyer flask

40 containing 40ml of nutrient solution S2 shaken at 240 rpm for

3 days at 25°C.

An intermediate culture was generated by aseptically transferring the preculture to 2L of nutrient solution S2 in a 3L fermenter. The fermenter was agitated at 500rpm, aerated at 0.5vvm, and the temperature controlled at 25°C for 3 days.

A production culture was generated by aseptically transferring an intermediate culture to a 75L fermenter containing 50L of nutrient solution P2. The production fermenter was stirred at 300rpm, aerated at 0.5vvm, and 10 temperature controlled at 25°C. During the production fermentation the pH was uncontrolled and remained between 5.5 and 6.5. In addition the dissolved oxygen tension remained above 80%. After 5 days incubation the fermentation was stopped and the culture was harvested.

The solutions used were as follows, percentages being by weight:

Nutrient Solution S2: 1.5% glycerol, 1.5% soya bean peptone, 1% D-glucose, 0.5% malt extract, 0.3% NaCl, 0.1% CaCO<sub>3</sub>, 0.1% Tween 80, 0.1% Junlon PW110 (Honeywell and Stein, Sutton, 20 Surrey, U.K.) adjusted to pH 6

Nutrient Solution P2: 3.65% sucrose, 1.20% glutamic acid (sodium salt), 0.02% K<sub>2</sub>HPO<sub>4</sub>, 0.98% MES, 0.05% KCl, 0.1% Tween 80, 0.002% MgSO<sub>4</sub>, 0.002% CaCl<sub>2</sub>, 2% vitamin mix solution (see below), 0.5% trace elements solution (see below), adjusted to pH 6

vitamin mix solution: 0.0025% thiamine, 0.0025% riboflavin, 0.0025% pantothenate, 0.0025% nicotinic acid, 0.0025% pyridoxine, 0.0025% thioctic acid, 0.00025% folic acid, 0.00025% biotin, 0.00025% cyanocobalamin, 0.00025% p-amino

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benzoic acid, 0.005% vitamin K, 0.2% Tween 80.

trace elements solution: 0.17% ZnSO<sub>4</sub>, 0.11% FeSO<sub>4</sub>, 0.02% MnSO<sub>4</sub>, 0.006% H<sub>3</sub>BO<sub>3</sub>, 0.012% CuSO<sub>4</sub>, 0.005% Na<sub>2</sub>MoO<sub>4</sub>, 0.005% CoCl<sub>2</sub>, 0.008% KI.

5

### Example 5

Extraction and purification of the phomalactone from Paecilomyces sp. 3527

The fermentation broth from Example 4 was harvested by

filtration and the clarified filtrate was divided equally into

aliquots and each aliquot was back extracted batch wise with

to fexame to remove non-polar impurities. The hexane

extracts were separated and discarded. Each batch of filtrate

was then back extracted with 2 x 8L of ethyl acetate. The

phomalactone and some impurities were extracted into the

solvent while many of the more polar impurities remained in

the aqueous phase. The ethyl acetate extracts were pooled and

concentrated to dryness under reduced pressure and redissolved

in 50ml of ethyl acetate.

Purification was achieved by normal phase chromatography using a Biotage Flash 75 chromatography system and a Flash 75 KP-Sil silica (32-62  $\mu$ m 60Å) column (internal diameter 7.5 x 30cm length) and an isocratic mobile phase (100% ethyl acetate, 200ml/min flow rate). 1L fractions were collected and analysed by thin layer chromatography using ethyl acetate as the developing solvent.

Phomalactone rich fractions (Rf 0.75) were pooled, evaporated to dryness under reduced pressure and subjected to further purification by preparative reversed phase HPLC using

a Beckman 350 HPLC with a Shandon Hyper prep HS BOS C18 (100 Å  $12\mu m$ ) column (internal diameter 10 x 30 cm length) and an isocratic mobile phase (85% water:15% acetonitrile, flow rate 170ml/min). Wavelength monitoring was at 210nm. The peak collected between 6-12 minutes was evaporated to dryness to yield the target phomalactone (6.0g).

Physicochemical data for the phomalactone are set out in Tables 7 to 9 below. Tables 8 and 9 show  $^1\mathrm{H}$  and  $^{13}\mathrm{C}$  NMR assignments respectively.

### 10 <u>Table 7</u>

	Phomalactone of formula (II)		
DCI-MS (m/z)	155 (MH).		
Molecular formula	C <sub>8</sub> H <sub>10</sub> O <sub>3</sub>		
UVA <sub>max</sub> nm	200		
IR (KBr) u cm <sup>-1</sup>	3425, 2923, 2855, 1716, 1629, 1381, 1262,		
·	1157, 1102, 1076, 1034, 969, 830.		

### Table 8

	Position	†δH/ppm in CDCl <sub>3</sub>
		Phomalactone of formula (II)
20	2	
	3	6.12 (1H, d, 9.6)
	4	6.97 (1H, dd, 9.7, 5.2)
	5	4.21 (1H, dd, 5.2, 3.2)
	6	4.85 (1H, m)
25	7	6.00 (1H, dqd, 15.3, 6.5, 1.0)
	8	5.75 (1H, ddq, 17.0, 6.9, 3.4)
100	9	1.82 (3H, ddd, 6.1, 1.0)

† The J values are in parenthesis (Hz)

Table 9

P	osition	δC/ppm in CDCl <sub>3</sub>	
		Phomalactone of formula (II)	
5	2	162.9	
	3	122.8	
	4	144.3	
	5	63.1	
li	6	80.9	
o	7	132.9	
	8	123.8	
i	9	17.8	

Synthesis of the 5,6-dihydro- $\alpha$ -pyrone of formula (I) wherein  $R^1$  and  $R^2$  are H and R is  $CH_3$  by esterification of the phomalactone with the fatty acid of formula (IIIa).

A solution of the phomalactone (20mg, 0.13mmol 1 eq) in 20 dry dichloromethane (2ml) was added to a stirred solution of the fatty acid of formula (III) (29mg, 0.13mmol), 1-(3dimethylaminopropyl) -3-ethylcarbodiimide hydrochloride (24.8mg, 0.13mmol, 1eq) and dimethylaminopyridine (1mg) in dry dichloromethane (5ml) at -78°C. The reaction mixture was 25 allowed to warm to room temperature and was stirred for a further 16 hours. The reaction mixture was washed with saturated ammonium chloride solution (3 X 5ml), water (1 X 5ml) and brine (1 X 5ml). The organic extract was dried over MgSO, filtered and concentrated in vacuo yielding a yellow 30 oil. Purification by preparative reversed phase HPLC yielded the 5,6-dihydro- $\alpha$ -pyrone of formula (I) wherein  $R^1$  and  $R^2$  are H and R is CH3 (11.1mg). HPLC, TLC, and 1H NMR analysis showed that this product was identical to the isolated natural product.

#### TNF- $\alpha$ Release from U937 Cells

The effect of the compounds of the invention on TNF- $\alpha$  5 release was investigated using a known method (Lozanski, G. et al., (1992), J. Rheumatol; 19, 921-26).

The human histolytic lymphoma U937 cell line was obtained from a commercial source (ECACC, Salisbury, UK) and maintained in RPMI 1640 medium supplemented with 2mM L-glutamine and 5% 10 fetal bovine serum. The cells were pretreated with 25 ng/ml PMA for six hours and then exposed to dose ranges of the compound to be tested followed by the addition of 1 ng/ml LPS.

After 18 hours incubation at 37°C with 5% CO<sub>2</sub> the cell

15 culture supernatants were harvested and stored at -70°C, until

required for determination of TNF-α secretion by Dissociation

Enhanced Lanthanide Fluorescence Immuno Assay (DELFIA). The

effect of the compounds on cell cytotoxicity was measured

using the tetrazolium salt, XTT (2,3-bis[2-methoxy-4-nitro-5
20 sulphophenyl]-2H-tetrazolium-5-carboxanilide salt) and effects

on protein synthesis were determined by investigation of [³H]

Leucine uptake.

In this test the compounds of the invention were found to inhibit TNF- $\alpha$  release at concentrations of from 10 to 0.1  $\mu$ M 25 (Table 10). At this concentration range, the compounds were not toxic and showed no effect on protein synthesis. The IC<sub>50</sub> ( $\mu$ M) values for LPS-induced TNF production in U937 cells for the 5,6-dihydro- $\alpha$ -pyrones for formula (I) are given in Table 11.

### IL-1 Release from Human Monocytes

The effect of the compounds of the invention on IL-Iβ

5 release was investigated using a known method (Bakouche, O. et
al., (1992), J. Immunol: 148, 84-91). Human monocytes were
purified by elutriation from Buffy coats obtained from normal
healthy donors after the separation of peripheral blood
mononuclear cells (PBMC) on Lymphoprep. The freshly isolated

10 monocytes were suspended in RPMI 1640 supplemented with 5% FBS
and exposed to dose ranges of the compound to be tested
followed by the addition of 1 ng/ml LPS. Cells were incubated
for 18 hours at 37°C with 5% CO<sub>2</sub> and the cell culture
supernatants harvested and stored at -70°C. Effects on the

15 production of IL-1β were determined using an ELISA.

The compounds of the invention were found to inhibit the release of IL-1 $\beta$  at concentrations of from 0.2 to 10  $\mu$ M (Table 12). At this concentration range the compounds were not toxic to monocytes. The IC<sub>50</sub> ( $\mu$ M) values for LPS-induced IL-1 $\beta$  production for the 5,6-dihydro- $\alpha$ -pyrones of formula (I) are given in Table 11.

Table 10: Inhibition of TNF-α release

	Concentration $\mu M$	% Inhibition of TNF- $\alpha$
R in formula (I) is CO <sub>2</sub> H	8 2 0.5 0.12	107 100 77 47
R in formula (I) is $\mathrm{CH}_3$	2.8 0.7 0.17 0.04 0.01	99 90 61 20 0.85

Table 11: IC<sub>50</sub>s for inhibition of LPS-induced cytokine

# 10 production by the 5.6-dihydro-α-pyrones of formula (I)

		IC <sub>50</sub> (μM)		
R <sup>1</sup> , R <sup>2</sup> in formula (I)	R in Formula (I)	LPS-induced TNF production in U937 cells	LPS-induced IL-β production (monocytes)	
н,н	CO <sub>2</sub> H	2	2	
н,н	CH <sub>3</sub>	0.08	0.19	
н, он	CO₂H	32	-	
он,н	CO₂H	31	-	

Table 12: Inhibition of the release of Il-1β

25		Concentration $\mu M$	% Inhibition of IL-1β
	R in formula (I) is CO₂H	20 2 0.5	99 81.5 14
	R in formula (I) is $CH_3$	5.5 1.1 0.22 0.044	88 72 41 23

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#### CLAIMS

1. A 5,6-dihydro- $\alpha$ -pyrone of formula (I)

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wherein R is  $CO_2H$  or  $CH_3$  and each of  $R^1$  and  $R^2$  is H; or R is  $CO_2H$ , one of  $R^1$  and  $R^2$  is H and the other is OH; or, when R is  $CO_2H$ , a pharmaceutically or veterinarily acceptable salt thereof.

- 2. A process for the preparation of a 5,6-dihydro- $\alpha$ -pyrone of formula (I) as defined in claim 1 or a pharmaceutically or veterinarily acceptable salt thereof, which process comprises:
- 20 (i) fermenting, in a source of carbon, nitrogen and inorganic salts, fungal strain *Phomopsis* sp. 22502 (CBS 313.96) or a mutant thereof which produces a said 5,6-dihydro- $\alpha$ -pyrone;
- (ii) isolating a said 5,6-dihydro- $\alpha$ -pyrone from the fermentation broth; and
  - (iii) if desired when the isolated said 5,6-dihydro- $\alpha$ -pyrone is the compound of formula (I) wherein R is  $CO_2H$ , converting the said 5,6-dihydro- $\alpha$ -pyrone into a pharmaceutically or veterinarily acceptable salt thereof.

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3. A process for the preparation of a 5,6-dihydro- $\alpha$ -pyrone of formula (I), as defined in claim 1, wherein R is  $CH_3$ , which process comprises esterifying the phomalactone of formula (II):

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with a fatty acid of formula (IIIa):

- 4. A pharmaceutical or veterinary composition
  20 comprising a pharmaceutically or veterinarily acceptable
  carrier or diluent and, as active ingredient, a compound as
  claimed in claim 1.
  - 5. A compound according to claim 1 for use in a method of treatment of the human or animal body by therapy.
- 25 6. A compound according to claim 5 for use as a cytokine production inhibitor.
  - 7. A compound according to claim 6 for use as an IL-1 production inhibitor.
    - 8. A compound according to claim 6 for use in the

treatment of an immunoinflammatory condition.

- A compound according to claim 8 for use in the treatment of rheumatoid arthritis, osteoarthritis, septic shock, psoriasis, atherosclerosis, inflammatory bowel disease,
   Crohn's disease or asthma.
  - 10. A compound according to claim 6 for use in the treatment of a central nervous system disorder.
  - 11. A process for the preparation of the phomalactone of formula (II) defined in claim 3, which process comprises:
- 10 (i) fermenting, in a source of carbon, nitrogen and inorganic salts, fungal strain *Paecilomyces* sp. 3527 (CBS 314.96) or a mutant thereof which produces the said phomalactone; and
- (ii) isolating the said phomalactone from the fermentationbroth.
  - 12. A fatty acid of formula (IIIb):

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wherein one of  $R^1$  and  $R^2$  is H and the other is H or OH.

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13. A process for the preparation of a fatty acid of formula (III):

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broth.

wherein one of  $R^1$  and  $R^2$  is H and the other is H or OH. which process comprises:

- (i) fermenting, in a source of carbon, nitrogen and inorganic salts, fungal strain *Phomopsis* sp. 22502 (CBS 313.96) or a mutant thereof which produces the said fatty acid; and
   (ii) isolating the said fatty acid from the fermentation
- 14. A biologically pure culture of fungal strain Phomopsis sp. 22502 (CBS 313.96) or a mutant thereof which
  15 produces a 5,6-dihydro-α-pyrone of formula (I) as defined in claim 1 or a fatty acid of formula (III) as defined in claim 13.
- 15. A biologically pure culture of fungal strain

  Paecilomyces sp. 3527 (CBS 314.96) or a mutant thereof which

  20 produces a phomalactone as defined in claim 3.
- sp. 22502 (CBS 313.96) or a mutant thereof as defined in claim 13, which process comprises fermenting strain *Phomopsis* sp. 22502 (CBS 313.96) or a said mutant thereof in a source of carbon, nitrogen and inorganic salts.
  - 17. A process for fermenting fungal strain *Paecilomyces* sp. 3527 (CBS 314.96) or a mutant thereof as defined in claim 14, which process comprises fermenting strain *Paecilomyces* sp.

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3527 (CBS 314.96) or a said mutant thereof in a source of carbon, nitrogen and inorganic salts.

# INTERNATIONAL SEARCH REPORT

Inte hal Application No PCT/GB 97/02907

	161740 37742307						
A. CLASSI IPC 6	FICATION OF SUBJECT MATTER C07D309/32 C12P7/40 C12P17/0 C12N1/14 //(C12N1/14,C12R1:649		57/03				
According to International Patent Classification (IPC) or to both national classification and IPC							
B. FIELDS SEARCHED							
Minimum documentation searched (classification system followed by classification symbols) IPC 6 C07D C12P C07C							
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched							
Electronic data base consulted during the international search (name of data base and, where practical, search terms used)							
C. DOCUMENTS CONSIDERED TO BE RELEVANT							
Category '	Citation of document, with Indication, where appropriate, of the rele	evant passages	Relevant to claim No.				
Α	P.WIPF ET AL.: "TOTAL SYNTHESEIS STRUCTURE ASSIGNMENT OF THE ANTIT ARANOROSIN."		1,12				
	JOURNAL OF ORGANIC CHEMISTRY., vol. 58, no. 25, 1993, EASTON US,						
	pages 7195-7203, XP002052631 see page 7195, paragraph EXPERIM page						
Х	7196						
^	A.MCKILLOP: "THE TOTAL SYNTHESIS OF THE 12 DIEPOXYCYCLOHEXANONEANTIBIOTIC ARANOROSIN" JOURNAL OF THE CHEMICAL SOCIETY, PERKIN TRANSACTIONS 1.,						
	vol. 12, 21 June 1996, LETCHWORTH GB, pages 1385-1392, XP002052632						
	see page 1385, paragraph EXPERIM						
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Further documents are listed in the continuation of box C.  Patent family members are listed in annex.							
<sup>o</sup> Special cal	tegories of cited documents :	"T" later document published after the inte					
"A" document defining the general state of the art which is not considered to be of particular relevance invention  "E" codic document but published on or offer the international."							
"E" earlier document but published on or after the international filing date "X" document of particular relevance; the claimed invention cannot be considered to cannot be considered to the cannot be considered to inventive step when the document is taken alone							
which is cited to establish the publication date of another "Y" document of particular relevance; the claimed invention							
"O" docume	cannot be considered to involve an inventive step when the  "O" document referring to an oral disclosure, use, exhibition or  other means  ments, such combination being obvious to a person skilled						
"P" document published prior to the international filing date but later than the priority date claimed in the art.  "&" document member of the same patent family							
Date of the actual completion of theinternational search  Date of mailing of the international search report							
20	O January 1998	04/02/1998					
Name and mailing address of the ISA  European Patent Office, P.B. 5818 Patentlaan 2  Authorized officer							
	NL – 2280 HV Rijswijk Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,	Francois. J					

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